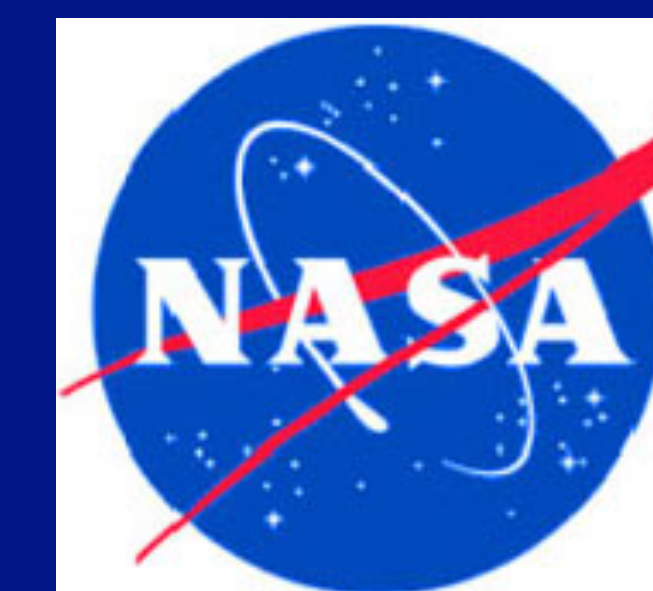




Evaluating Effects of H₂O and Overhead O₃ on Global Mean Tropospheric OH Concentration



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1. Scientific motivation for studying tropospheric OH

Sensitivity of OH^{TROP} to climate change

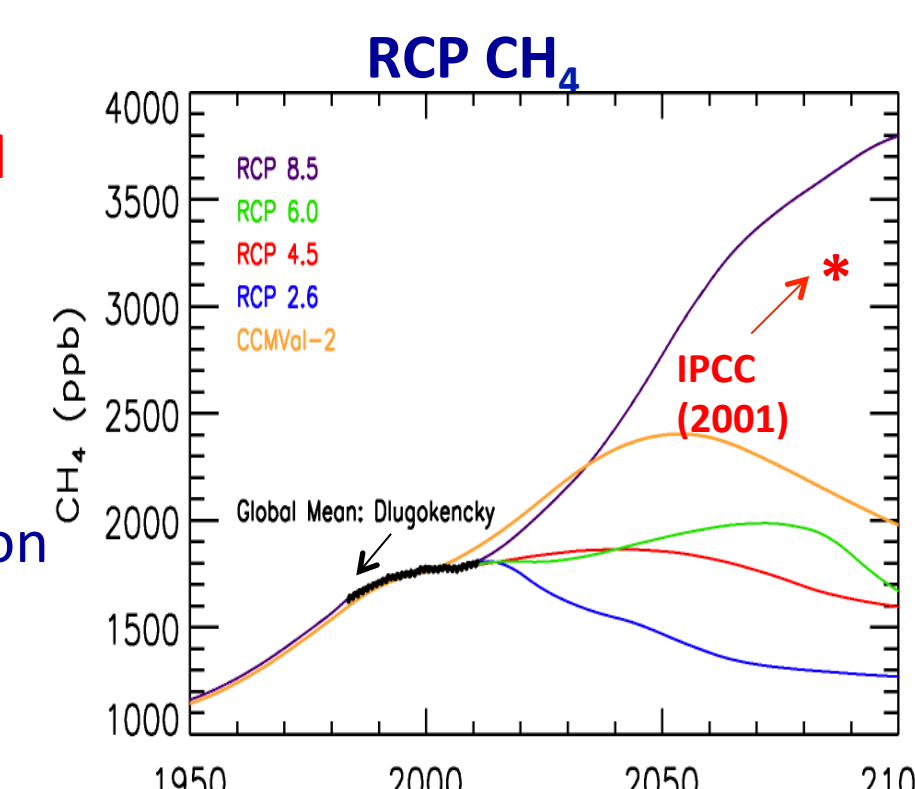
- IPCC (2001) predicted future OH would fall due to rising CH₄
- Best knowledge of global tropospheric OH (OH^{TROP}) comes from analysis of CH₃CCl₃ observations
- The CH₄ lifetime inferred from OH^{TROP} based on CH₃CCl₃ is 8.9 years; yet many sources use a CH₄ lifetime of ~12 years, which was found by modeling studies described in IPCC (2001)
- The 12 year CH₄ lifetime is central to subsequent IPCC reports and is called the “perturbation lifetime”
- Our work is motivated by understanding:
 - what factors other than rising CH₄ will affect OH^{TROP}
 - veracity of notion that future OH^{TROP} will decline

Importance of OH^{TROP}

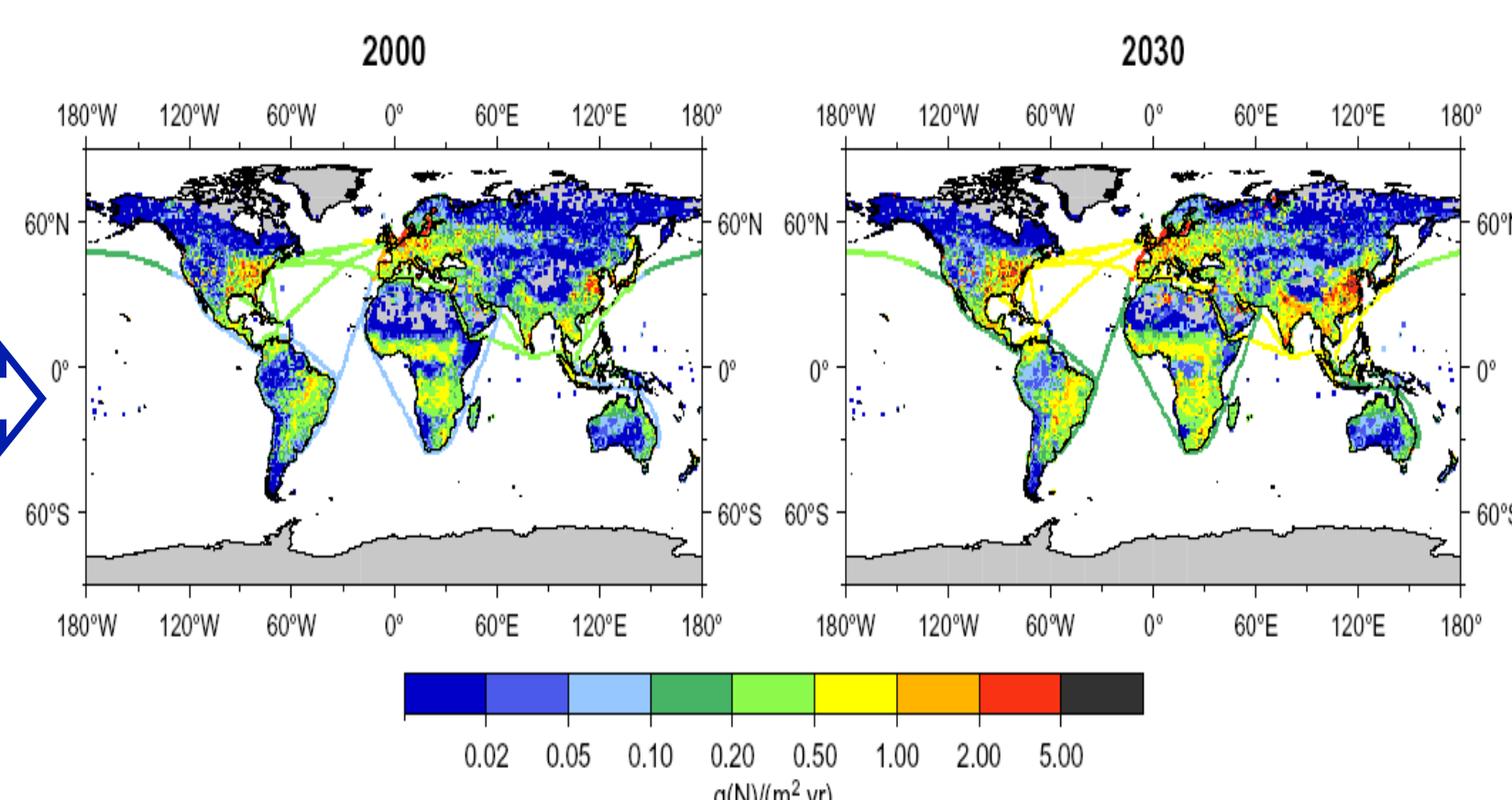
- OH serves as the main sink for many species: CO, CH₄, CH₃CCl₃, all HCFCs and HFCs, biogenic VOCs (isoprene and terpene), anthropogenic VOCs (formaldehyde, benzene, toluene, etc.)
- HO_x is integral to O₃ formation and destruction pathways; alone, HO_x depletes O₃, but with NO_x and VOCs, HO_x creates O₃

Methane Oxidation and OH

- Under low NO_x, the oxidation of CH₄ destroys HO_x
- Under high NO_x, the oxidation of CH₄ produces HO_x
- **Loss of methylperoxy radical CH₃O₂ with NO or HO₂ is critical**
- Future levels of CH₄ and NO_x are highly uncertain
 - CH₄ between now and 2100 varies dramatically among RCP scenarios
 - NO_x between now and 2100 will depend on whether the developing world implements selective catalytic reduction on coal power plants and catalytic converters on cars
- Future OH^{TROP} will also depend on overhead column O₃, local humidity, and biogenic emission of VOCs, etc.
- **Given this complexity, we expect CCMs to project a wide range of values for OH^{TROP} during the rest of this century**



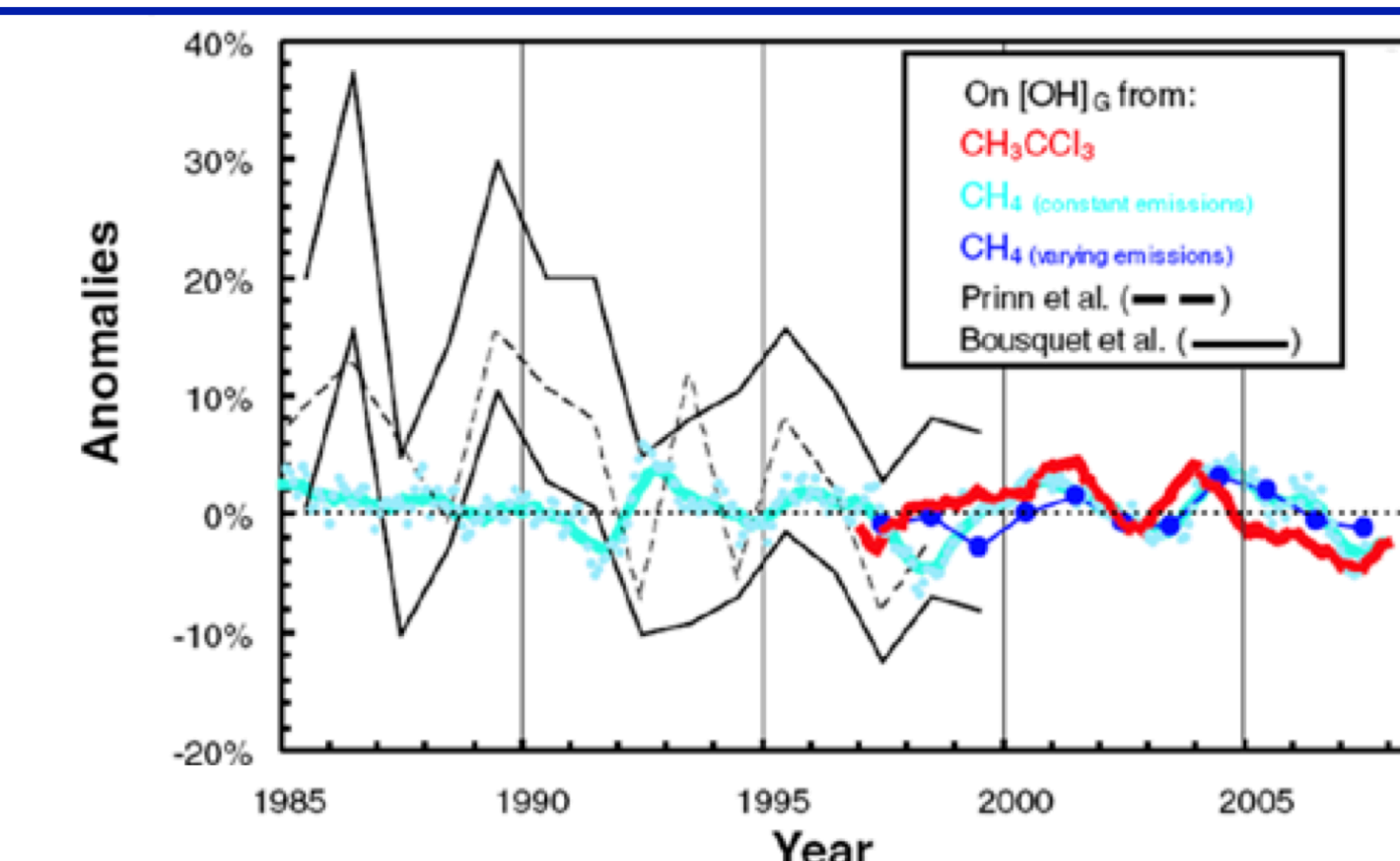
NO_x emissions from industry, power generation, traffic, domestic heating, and biomass burning used as input for prior CCM calculations, for years 2000 (38.0 Tg N / yr total) and 2030 (67.6 Tg N / yr total). From Eyring et al., ACP, 2007.



3. Preliminary results for expected changes in OH^{TROP}

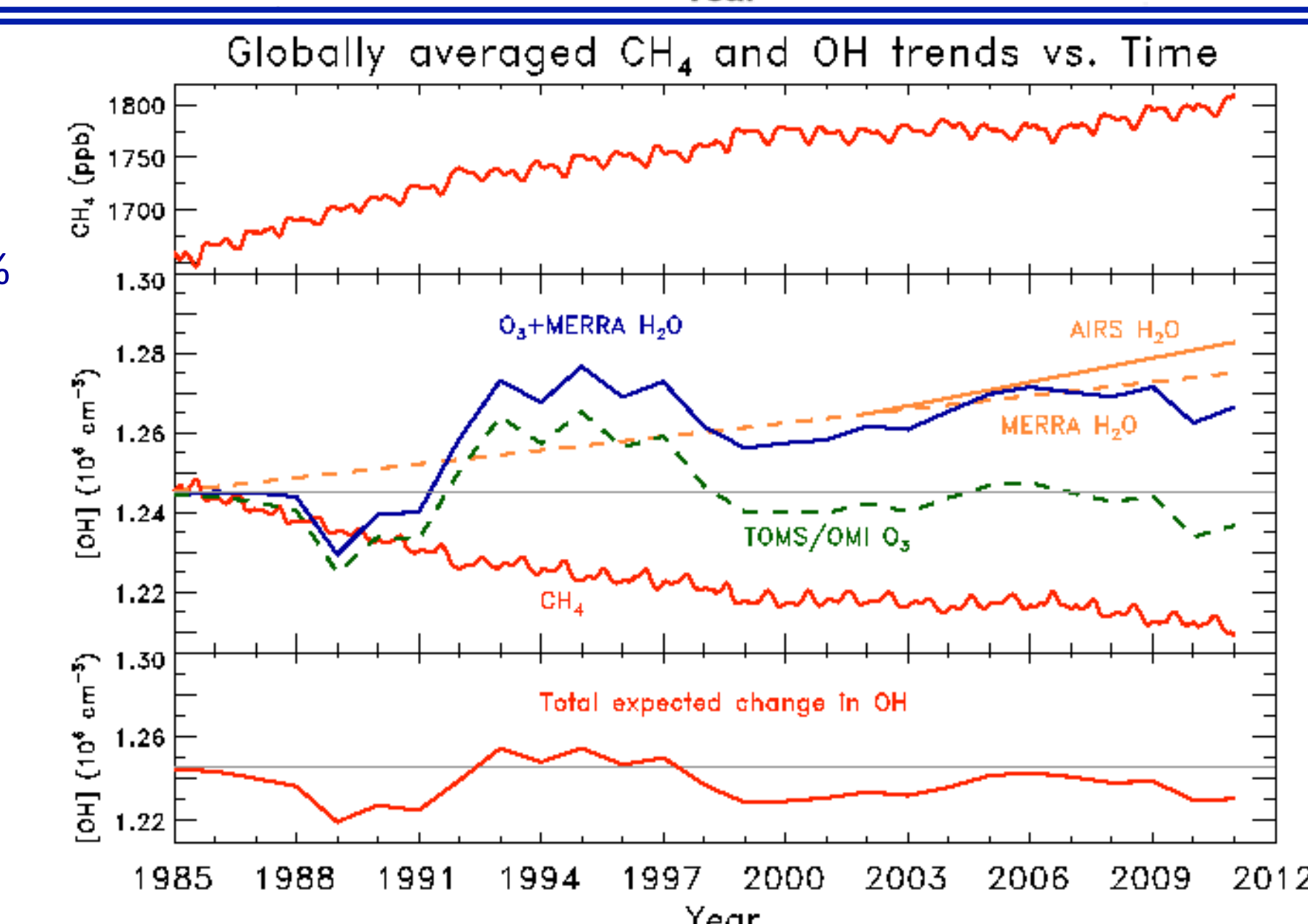
Previous Work on OH^{TROP} Trends

- Montzka *et al.*, 2011 found that OH^{TROP} does not vary interannually (from 1997 to present)
- Prior studies by Prinn *et al.*, 2001 and Bousquet *et al.*, 2005 suggest large interannual variability in OH^{TROP} (1985-2000)
- We suggest the OH^{TROP} behavior in all three studies may be physically possible, based on our preliminary results



Current Conclusions from OH^{TROP} Analysis

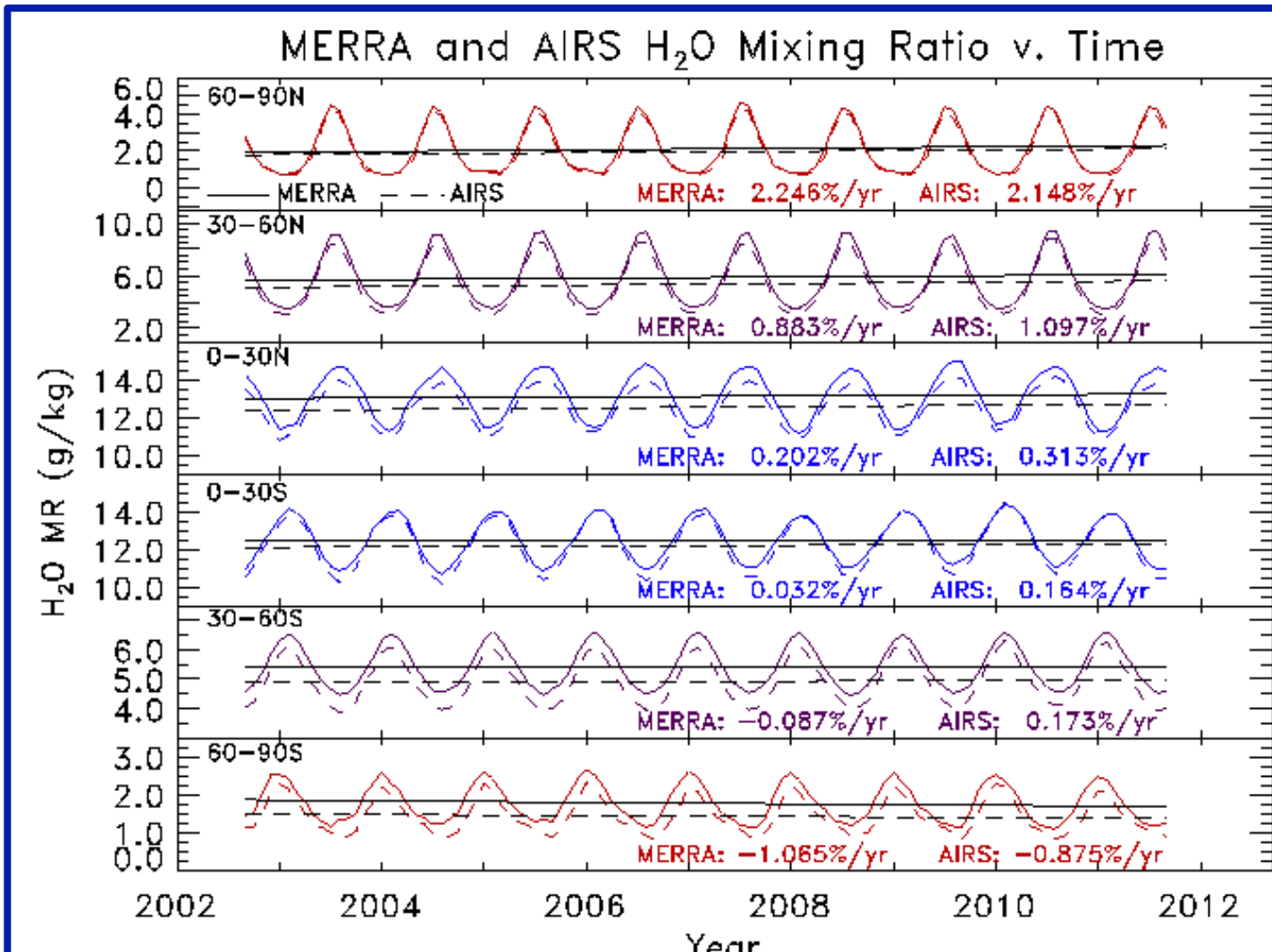
- Effect of CH₄ on OH^{TROP} is taken from IPCC (2001), Section 4.2.1.1, which states that “the feedback of CH₄ on tropospheric OH” found using contemporary chemical transport models is -0.32% for every 1% increase in CH₄ (red line, middle panel of figure to right)
- Primary effect of overhead O₃ is rise in OH^{TROP} following the 1991 eruption of Mount Pinatubo (green dashed line) due to enhanced removal of stratospheric O₃ by volcanic aerosol
- Rising H₂O from MERRA and AIRS increases OH^{TROP} (orange lines) by an amount comparable to the decrease expected from rising CH₄
- Overall expected change in OH^{TROP} (bottom panel) shows higher level of interannual variability prior to ~1999 and lower variability thereafter



2. Methods for estimating changes in OH^{TROP}

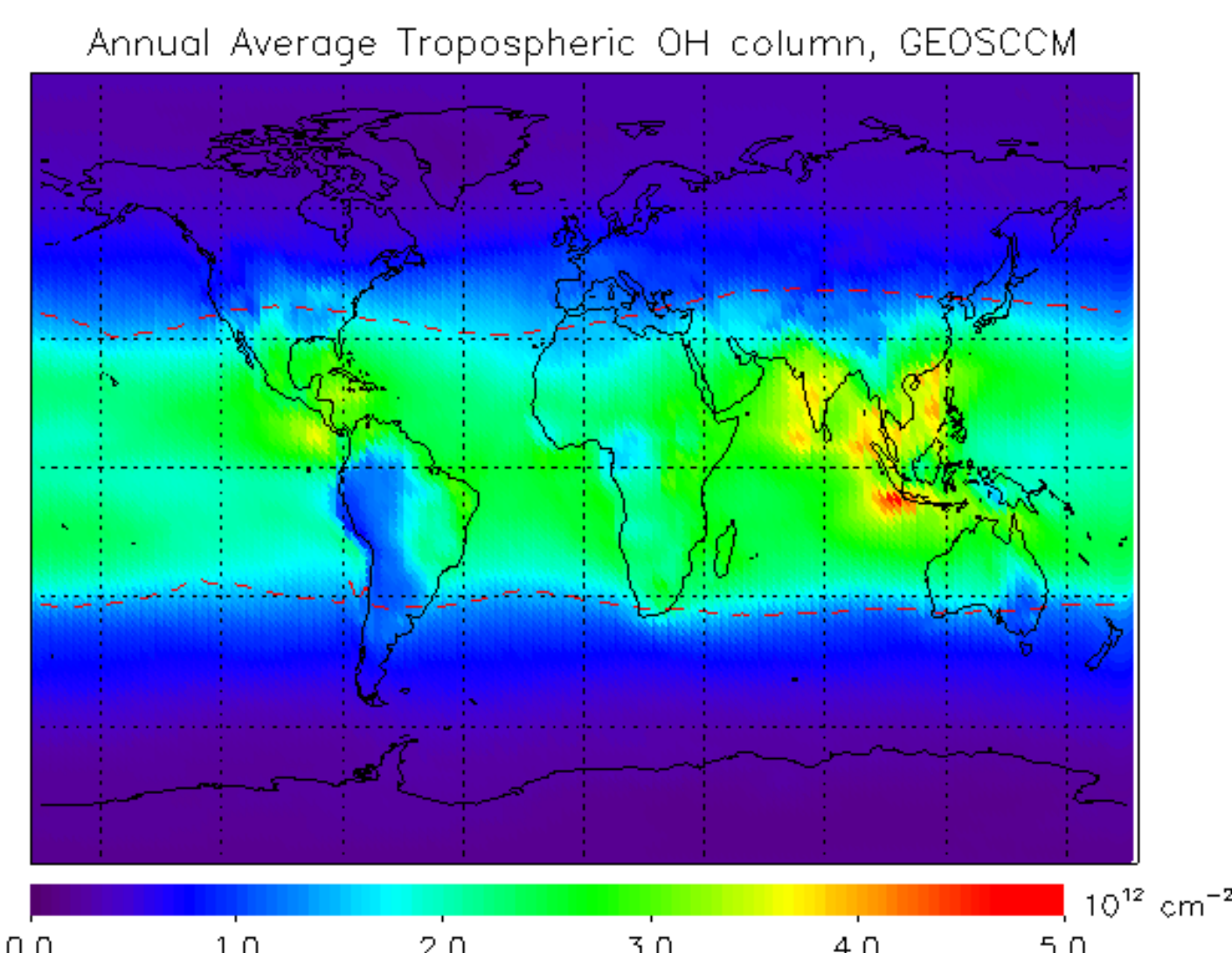
Base Case OH Conditions

- Initial modern-day conditions for OH are taken from a time-slice run of GEOS CCM using 2005 emissions
- Monthly mean mixing ratios of OH and related species are provided on a 144 longitude, 91 latitude, 72 pressure level grid
- Calculated changes in OH due to H₂O and overhead O₃ are applied to initial OH field



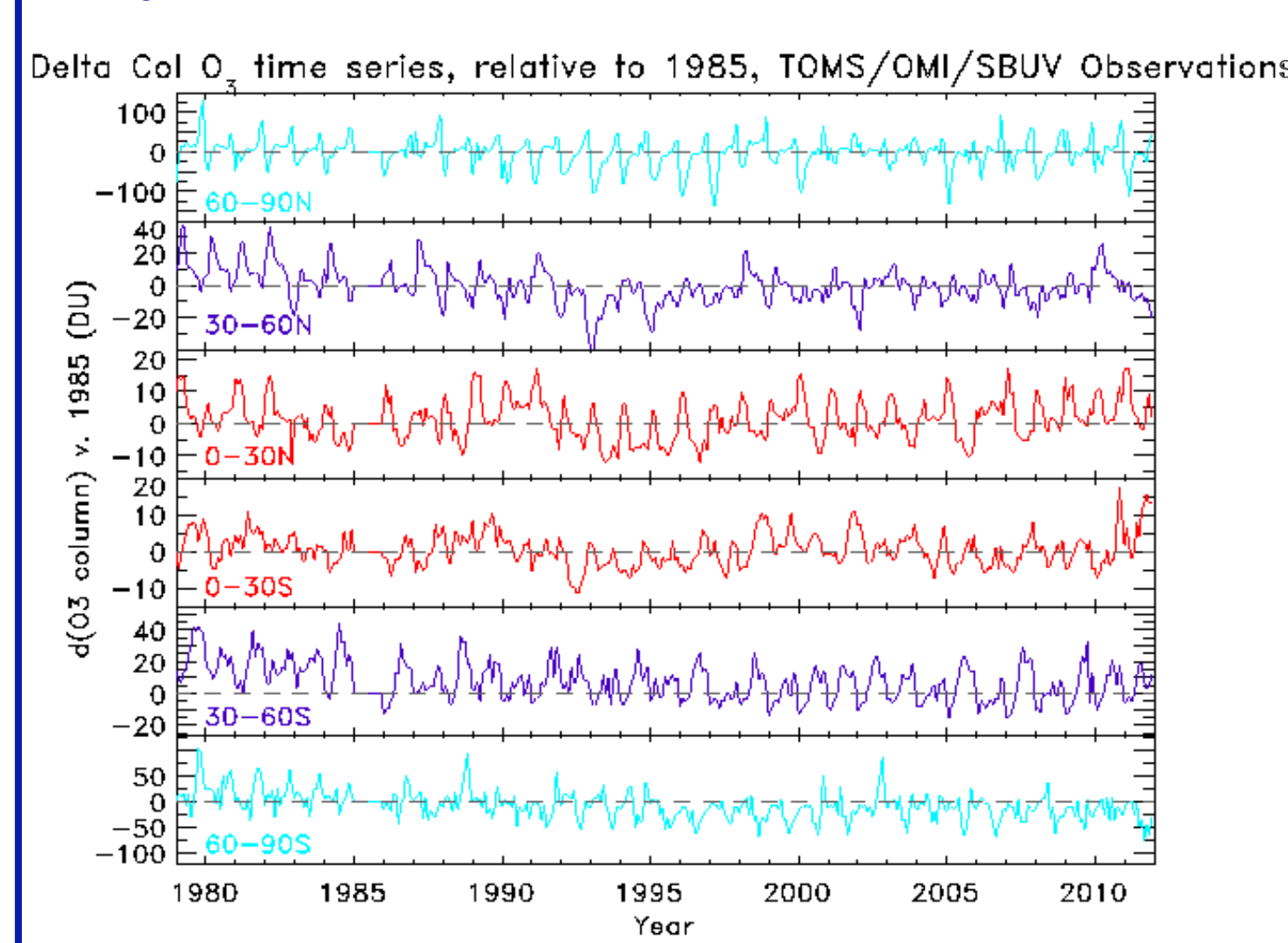
MERRA & AIRS H₂O

- Specific humidity files from the NASA Modern-Era Retrospective analysis for Research and Applications (MERRA) starting prior to 1985, and
- H₂O mixing ratio files from the NASA Atmospheric Infrared Sounder (AIRS) starting in 2002 were used to establish trends in H₂O by latitude
- OH^{TROP} is assumed to follow the square root of the change in H₂O using a steady-state assumption
- We plan to refine the effect of changing H₂O on OH^{TROP} by examining reaction rates from archived runs of GEOS CCM



Overhead O₃ Observations

- Total column O₃ trends were obtained from the NASA merged O₃ data set, consisting of measurements from SBUV, TOMS, and Aura OMI instruments
- We then use our photolysis code to estimate the impact on J(O₃) → O(¹D) of decreasing initial GEOS CCM overhead O₃ columns by amount suggested by the NASA product
- OH^{TROP} is assumed to change by the square root of J(O₃) → O(¹D)



4. Future plans for refining estimates of ΔOH^{TROP}

Improve Estimate of d(OH^{TROP}) / d(H₂O)

- Reaction rates from recent runs of GEOS CCM are archived for reactions such as:
$$\text{H}_2\text{O} + \text{O}(^1\text{D}) \rightarrow 2\text{OH}$$
- Using these reaction rates we will determine the proportion of OH that is produced via reaction with H₂O
- The determined scaling factor would be used to calculate a new ΔOH^{TROP} based on the H₂O trends
- Estimate time- and pressure-varying values of d(OH^{TROP}) / d(H₂O)
- Evaluate discrepancies between MERRA and AIRS H₂O trends

Propagate Uncertainties

- Calculate uncertainties in AIRS and MERRA H₂O and NASA O₃ product

Evaluate CH₄/OH Feedback

- We will use a box model (details below) to probe relationship between CH₄ and OH^{TROP} and its dependence on NO_x

Box Model

- We will use the GSFC Combined Stratosphere-Troposphere (COMBO) box model provided by Chang Lang (JHU):
- GMI chemical mechanism
 - 118 species, 321 thermal reactions, and 81 photolysis reactions
 - 5 modules:
 - Aerosol optical depth & surface area
 - Photolysis scheme
 - Thermal reactions scheme
 - Differential eqn solver
 - Input-output
 - Fast-JX photolysis & SMVGear II solver

- Evaluate standard deviation in average fraction of OH production occurring via H₂O + O(¹D)
- Estimate uncertainty in the box model evaluation of d(OH^{TROP}) / d(CH₄)

Assessing OH^{TROP} in CCMs

- Through our involvement with the IGAC / SPARC Chemistry-Climate Model Initiative, we have requested:
 1. hourly, instantaneous output from participating CCMs 1 day/season, 1 year/decade
 2. archival of all species, reaction rates, J-values, and physical parameters relevant to OH chemistry
 3. this output for both the REF-C1 (hindcast) and REF-C2 (future) runs
- We plan to assess the causes of differences between OH in the CCMs
- Use of the box model enables us to distinguish between OH differences due to **chemical mechanism** and those due to **differences in OH precursors**
- We can also use this output to predict future trends in OH^{TROP}, based on CH₄, H₂O, and overhead O₃ from the future CCM runs